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molarity (M). Molarity is defined as the number of moles of solute (mol) divided by the total volume (V) of the solution in liters (L).

Molarity = moles of solute per liter of solution ($M = \text{mol} / \text{L}$).

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of the new solution? (Answer:
9-801 HN03) = ML (0.8M -N
3-30.00 ml, of water is added to
20.00 ml, of 0.0500 M sodium
carbonate. What is the molarity of
sodium carbonate in the new
solution? (Answer: 0.0200 M)

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(20.00 m C (350. 00 casF) 4.

What volume of 0.100 M NaCl is needed to make 5.00 ml, of 0.200 M solution? (Answer: 0.01 (0.100 M) V)

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