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formula is  $C_4H_{10}O$

2. A was gone under  
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and listed as below. a gas was emitted when sodium was added to the A. A does not react with with NaOH.

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the following statements about an  $sp$  hybridized carbon is FALSE? a. It is divalent. b. It forms bonds that are linear. c. It has two  $p$  orbitals. d.

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chiral carbon. The answer is (B) (by definition) 4. The specific rotation of pure (R)-2-butanol is  $-13.5^\circ$ . What % of a mixture of the two enantiomeric forms is (S)-2-butanol if the specific rotation of this mixture is  $-5.4^\circ$  ?  
(A) 40%(B) 30%(C) 60%(D) 70%(E) None of

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the above The answer is  
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favoring R.

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