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Chapter 4 - Atoms and elements

Chapter 4 - Reactions in Aqueous Solution: Part 1 of 6

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See Section 4.1 (Page 103) . ANSWER: Correct Element Names, Symbols, and Atomic Numbers
Learning Goal: To identify elements by their chemical symbol and their atomic number.
Chemical elements are substances that are composed of only one kind of atom and cannot be broken down into simpler substances.

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Mastering Chemistry - Chapter 4. For soluble salts, the molarity of a particular ion is equal to the molarity of that compound times the subscript for that ion. For example, 1 M of AlCl_3 is 1 M in Al^{3+} and 3 M in Cl^- . 1 M of $(\text{NH}_4)_2\text{SO}_4$ is 2 M in NH_4^+ and 1 M in

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