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And

1. Study the flow chart

below and answer the

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questions that follow: 4.

Name reagent Z. 5.

Describe the process
which takes place in

step 2. 6. Identify the
white solid. 2. a)

Starting from solid
magnesium oxide,
describe how a solid
sample of magnesium
hydroxide can be
prepared. b) Give one
use of magnesium
hydroxide. 3.

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INSTRUCTIONS

Attempt all the question
in the question in the
spaces provided All the

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working must be clearly shown where appropriate. Negligence will be penalized.

QUESTIONS 1. The table below shows some information concerning partials of certain elements. Element
Atomic number Number of electrons Mass
Number P 19 18 39

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1.1 The structure of the

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atom. When scientists started exploring matter, they realised that matter can be divided into smaller and still smaller particles. They called the smallest particle an 'atom'. The name 'atom' was derived from the Greek word 'atomos', meaning 'indivisible'.

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Answer all the questions
in the spaces provided.

2. Mathematical tables
and electronic
calculators may be used.

3. All working must be
clearly shown where
necessary 1. The table
below shows the pH
values of solutions J to

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1. What is the first element on the periodic table? 2. What is the centre of an atom called? 3. True or false? Acids have a pH level below 7. 4. What is the main gas found in the air we breathe?

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2) The modern atomic mass unit is based on _____. Answer:

Carbon-12.

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Movement of substances from one part of the body to another.
- ii) Explain the necessity of transport in plants and animals. make nutrients move from one point to another.
movement of respiratory gases i.e. oxygen and carbon IV oxide.
elimination of metabolic

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