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2 MATTER and Its
Properties Matter -
Anything that has
MASS and VOLUME.

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ATOMS and
MOLECULES the
basic building
blocks of matter.

ATOM-- the
smallest unit of an
element that still
maintains the
properties that
element.

ELEMENT- a pure
substance made up
of only one type of
atom.

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Solving Lab 1.
Explain why the
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compressed gas
must be controlled
for practical and
safe use.

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Reviewing

Vocabulary Match

the definition in
Column A with the
term in Column B.

Column A Column

B 1. The set of
frequencies of the
electromagnetic
waves emitted by
the atoms of an
element 2. The

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Change. extensive

property. intensive

property. extensive

properties

(examples)

intensive property

(examples) a

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property that depends on the amount of matter in a sample. a property that depends on the type of matter in a sample not.... mass, volume, length, weight.

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number of moles of
known substance C

H C H C H O O O

CO CO CO H O H O

H O number of

moles of unknown
substance O CO H

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O C H C O H O C H

O H O C H O C O

use mole ratio 5

mol O 2 1 mol C 3

H 8 3 mol CO 2 1

mol C 3 H 8 4 mol

H 2 O 1 mol C 3 H

8 1 mol C 3 H 8 5

mol O 2

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change" is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often

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evidence of a
chemical change.
Properties of
Compounds Quite
different properties
than their
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elements.

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