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Answers 1. Circle the molecules in the following list. H₂ HCl P₄ H₂O Fe S₈ CO₂ Ne 2. Circle the compounds in the following list. H₂ HCl P₄ H₂O Fe S₈ CO₂ Ne 3. Label the

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Answers as ionic or
covalent: HCl - covalent P4
- covalent H2O - covalent
NaCl - ionic Ca3(PO4)2 -
ionic CO2 - covalent BaS -
ionic 4. Predict the charge
each element below would
prefer to have ...

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molecules in the following
list. H₂ HCl P₄ H₂O Fe S₈
CO₂ Ne 2. Circle the
compounds in the following

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Answers. H₂ HCl P₄ H₂O Fe S₈

CO₂ Ne 3. Label the

following as ionic or

covalent: HCl - covalent P₄

- covalent H₂O - covalent

NaCl - ionic Ca₃(PO₄)₂ -

ionic Page 4/8. Download

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Homework Answers CO2 -
covalent BaS - ionic 4 ...

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Answers 2 CHEM 1411. Chapter
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(Homework) 11. The formula for sodium sulfide is A) NaS. B) K_2S . C) Na_2S . D) Na_2S . E) SeS. 12. Almost all the mass of an atom is concentrated in the nucleus.

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(Homework) work on the
"click and go". Each problem
has to be repeated 5 Page
6/8. Where To ...

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Answers
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Na 2 S. E) SeS. 12. Almost
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"click and go ...

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PLAY. John Dalton. 1.
elements composed of
extremely small particles
called atoms; 2. atoms of

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Answers
one element are different from atoms of all other elements 3. compounds are composed of atoms of more than one element; 4. chemical reaction involves only the separation, combinations, or

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Answers
rearrangement of atoms; not
creation or destruction .
law of definite proportions
...

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atoms. 2. atoms of one
element cannot be converted

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Answers
into atoms of another element in a chemical reaction. 3.all atoms of an element have the same number of protons and electrons which determines the chemical behavior of the element.

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illustrates the concept of
potential energy and which
illustrates kinetic energy?

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(a) water stored in a water tower. (b) A rubber band stretched around a newspaper. (c) A basketball dropping through a net. (d) A car driving down the highway. (e) A flashlight battery. 2. Which of the

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Answers following processes are
exothermic? Endothermic? (a
...

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8. Molecular Geometry and
Bonding Theories (Homework)
W Multiple Choice Identify
the choice that best
completes the statement or
answers the question. _____

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Answers
1. The electrons in the outer shell of an atom are involved in bonding. Another name for the ...

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for sodium sulfide is A)
NaS. B) K_2S . C) NaS_2 . D)
 Na_2S . E) SeS . 12. Almost
all the mass of an atom is

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Answers concentrated in the nucleus.
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(Homework) work on the
"click and go". Each problem
has to be repeated 5 times,
to practice. Copy ...

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Following Equations: A. AI

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(s) + H₂SO₄ (aq) ...

Question: CHEM 1411

Worksheet- Chapter 7 Name 1.

Balance The Following

Equations: A. Al (s) + H₂SO₄
(aq) → Al₂(SO₄)₃ (aq) + H₂

(g) B. C₂H₆ (g) + O₂ (g)

CO₂ (g) → H₂O (l) C. Pb(NO₂

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'covalent bond' on the
chemical community and the
great novelty of the idea
that was conceived 11 years
before quantum mechanics was
able to offer the mechanism
of electron pairing and

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